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ID: A. 1. CHEMISTRY: Chapter 10 Prep-Test. Matching. Match each item with the correct statement below.
a. calorimeter d. temperature b. calorie e. specific heat c. joule f. heat ____ 1. quantity of heat needed to raise the temperature of 1 g of water by 1 C ____ 2. SI unit of energy ____ 3. energy transferred between 2 objects because of temperature difference ____ 4.

CHEMISTRY: Chapter 10 Prep-Test

Review of Chemistry of Matter (Chapter 2, & 3 Test Review)ANSWERS Directions: Review and check all your answer with this answer sheet to get ready for the test on Chemistry of Matter (Chapter 2 & 3) Chemistry chapter 10 test review answers.

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space

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provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

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Chemistry in the Laboratory (7th Ed) Postma, Roberts, Hollenberg (ISBN 978-1-4292-1954-9 Lab Notebook

CHEMISTRY 101

33) Name any two nuclei used to make an atom bomb: Answer: Uranium, Plutonium. 34) A gaseous mixture contains oxygen and nitrogen in the ratio of 1: 4 by weight. Therefore the ratio of their number of molecules is. (a) 1: 4 (b) 1: 8 (c) 7: 32 (d) 3: 16. Answer: C.

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