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SHORT ANSWER Answer the following questions in the space provided. 1. Match the four compounds on the right to their descriptions on the left. b an ionic compound that is quite soluble in water (a) HCl c an ionic compound that is not very soluble in water (b) NaNO<sub>3</sub>

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b ...

## 6 Chemical Bonding

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Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. \_\_\_\_\_ 7. A solution containing 35 g of Li<sub>2</sub>SO<sub>4</sub> dissolved in 100 g of water is heated from 10 ° C to 90 ° C. According to information in the figure, this

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temperature change would result in a. an additional 5 g of Li<sub>2</sub>SO<sub>4</sub> in ...

Assessment Chapter Test A - Ed W. Clark High School

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